NEW VERTEX-EDGE SOMBOR, NIRMALA AND MISBALANCE INDICES

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ABSTRACT. The molecular structure of a compound contains all the information which would help to determine its chemical, biological, and physical properties. With the help of a theoretical descriptor tool known as topological indices, one can assess these properties. In this paper, we define new vertex-edge degree topological indices namely, the *ve*-degree Sombor index (SO_{ve}) , the *ve*-degree Nirmala index (N_{ve}) , and the *ve*degree Misbalance prodeg index (MPI_{ve}) . The chemical applicability of those indices have been studied and found good correlation coefficient with different physical/chemical properties of octane isomers. Further, we found the values for the standard graphs and the bounds for SO_{ve} and N_{ve} in terms of $MPI_{ve}, M'_{\beta ve}(G)$, and $F_{ve}(G)$.

Keywords: vertex-edge degree; topological indices.

AMS Subject Classification: 05C07.

1. INTRODUCTION

Graph theory played a significant role in molecular chemistry, robotics, physics, networks computer science, statistics, biological activities, and data science. A topological index is a unique number that is mathematically derived from the graph structure. In theoretical chemistry, many such topological indices have been considered, and have more applications in a quantitative structure-property relationship (QSPR) and quantitative structure-activity relationship (QSAR).

The (QSPR)/(QSAR) studies have an important role in material sciences [1, 2, 3]. The vertex-edge topological indices is a new idea and recently gaining more interest in applied sciences [4, 5, 6, 7, 8, 9, 10, 11, 12]. Let G = (V, E) be a simple connected graph. The number of edges that are incident with the vertex u is known as the degree of the vertex u and is denoted by, d(u). In [5], the set $N(u) = \{u \in V(G) : uw \in E(G)\}$ and $N[u] = N(u) \cup \{u\}$ are called as open and closed neighbourhood of the vertex u. The number of different edges that are incident to any vertex from N[u], denoted by $d_{ve}(u)$ and called as ve-degree.

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The first ve-degree Zagreb $\alpha(M^1_{\alpha ve})$ index, the first ve-degree Zagreb $(M^1_{\beta ve})$ index, the second ve-degree Zagreb (M_{ve}^2) index, ve-degree Randic (R_{ve}) index, the ve-degree atombond connectivity (ABC_{ve}) index, the ve-degree geometric-arithmetic (GA_{ve}) index, the ve-degree harmonic (H_{ve}) index, the ve-degree sum-connectivity (χ_{ve}) index, and veforgotten index are defined as,

$$\sum_{u \in V} d_{ve}(u)^2, \quad \sum_{uw \in E} (d_{ve}(u) + d_{ve}(w)), \quad \sum_{uw \in E} (d_{ve}(u) \times d_{ve}(w)), \quad \sum_{uw \in E} (d_{ve}(u) \times d_{ve}(w))^{-\frac{1}{2}}, \\ \sum_{uw \in E} \left(\frac{d_{ve}(u) + d_{ve}(w) - 2}{d_{ve}(u) \times d_{ve}(w)}\right)^{\frac{1}{2}}, \quad \sum_{uw \in E} \frac{2(d_{ve}(u) \times d_{ve}(w))^{\frac{1}{2}}}{d_{ve}(u) + d_{ve}(w)}, \quad \sum_{uw \in E} (d_{ve}(u) + d_{ve}(w))^{-\frac{1}{2}}, \quad \text{and} \quad \sum_{uw \in E} (d_{ve}(u)^2 + d_{ve}(w)^2)$$

respectively. Recently new topological indices have been defined, Sombor index (SO(G)) [13], Nirmala index N(G) [14], and Misbalance prodeg index MPI(G) [15].

Some properties of Sombor and Nirmala indices have studied in [15, 16]. In [17, 18] different version of Sombor index is studied and application found in [19]. In [20] the ve-degree Sombor index (SO_{ve}) is defined as, $SO_{ve} = \sum_{uw \in E} \sqrt{d_{ve}(u)^2 + d_{ve}(w)^2}$. Further, in this paper we define, the ve-degree Nirmala index (N_{ve}) , and the ve-degree Misbalance prodeg index (MPI_{ve}) as follows:

$$N_{ve} = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)}, \ MPI_{ve} = \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right)$$

respectively. In the next section, we will discuss the chemical applicability of the SO_{ve} , N_{ve} , and MPI_{ve} .

2. Chemical Significance of the SO_{ve} , N_{ve} , and MPI_{ve}

Here, we compute the SO_{ve} , N_{ve} , and MPI_{ve} of octane isomers and molecular graph of octane isomers are shown in Figure 1. We investigate the predictive power of the SO_{ve} , N_{ve} , and MPI_{ve} for certain Physico-chemical properties mainly, acentric factor (AcenFac), entropy (S), enthalpy of vaporization (HVAP), and standard enthalpy of vaporization (DHVAP) of octane isomers. In Table 1, the values of Acentric factor, Entropy, Enthalpy of vaporization (HVAP), Standard enthalpy of vaporization (DHVAP), SO_{ve} , N_{ve} , and MPI_{ve} for octane isomers are tabulated. The correlation between ve-degree topological indices and many physicochemical properties of octane isomers is found in Table 2. And it is noted that all indices show a negative strong correlation, therefore these graph invariants are compared with each other by using squares of the correlation coefficients. The graphical representation of the highest values of correlation between ve-degree topological indices are shown in Figure 2.

3. Results of SO_{ve} , N_{ve} , and MPI_{ve} on some stranded Graphs.

Here, we found the values of SO_{ve} , N_{ve} , and MPI_{ve} for the particular graph.

Proposition 3.1. Let $K_{m,n}$ be a complete bipartite graph. Then

i. $SO_{ve}(K_{m,n}) = \sqrt{2}(mn)^2$

ii.
$$N_{ma}(K_{m,n}) = mn\sqrt{2mn}$$

ii. $N_{ve}(K_{m,n}) = mn\sqrt{2mn}$ iii. $MPI_{ev}(K_{m,n}) = 2mn(\sqrt{mn}).$

Proof. Let $K_{m,n}$ be a complete bipartite graph with m + n vertics and $|V_1| = m$, $|V_2| = n$, $V(K_{m,n}) = V_1 \cup V_2$. Clearly, every vertex of $v \in V_1$ has mn different edges that incident to any vertex form N[v] and every vertex of $u \in V_2$ has mn different edges that incident to any vertex form N[u].

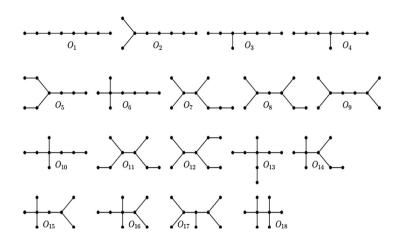
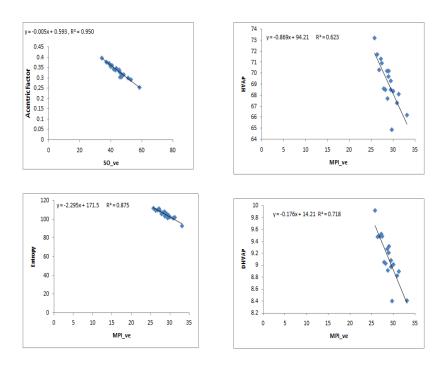
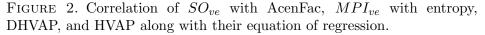


FIGURE 1. Graphs of Octane isomers.





i.

$$SO_{ve}(K_{m,n}) = \sum_{uw \in E} \sqrt{d_{ve}^2(u) + d_{ve}^2(w)} = \sum_{uw \in E} \sqrt{(mn)^2 + (mn)^2} = \sqrt{2}(mn)^2$$

$$N_{ve}(K_{m,n}) = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)} = \sum_{uw \in E} \sqrt{mn + mn} = mn\sqrt{2mn}$$

iii.

ii.

$$MPI_{ve}(K_{m,n}) = \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right) = \sum_{uw \in E} \left(\sqrt{mn} + \sqrt{mn} \right) = 2mn \left(\sqrt{mn} \right).$$

| Molecule | AcenFac | Entrony | HVAP | DHVAP | SO_{ve} | N_{ve} | MPIve |
|-----------------------------|----------|---------|-------|-------|-----------|----------|--------|
| | 0.255294 | 93.06 | 66.2 | 8.41 | | 23.641 | 33.166 |
| 2,2,3,3-Tretramethyl-butane | 0.200294 | 95.00 | | 0.41 | | | 55.100 |
| 2,4-Dimethyl-hexane | 0.344223 | 106.98 | 68.5 | 9.029 | 41.728 | 20.048 | 28.218 |
| 2-Methyl-heptane | 0.377916 | 109.84 | 70.3 | 9.484 | 37.069 | 19.594 | 26.815 |
| 2-Methyl-3-ethyl-pentane | 0.332433 | 106.06 | 69.7 | 9.209 | 45.333 | 20.653 | 28.938 |
| 3-Ethyl-hexane | 0.362472 | 109.43 | 71.7 | 9.476 | 40.613 | 19.605 | 26.356 |
| 2,2-Dimethyl-hexane | 0.339426 | 103.42 | 67.7 | 8.915 | 42.836 | 20.361 | 28.722 |
| 3-Methyl-heptane | 0.371002 | 111.26 | 71.3 | 9.521 | 38.786 | 19.322 | 27.205 |
| 2,3-Dimethyl-hexane | 0.348247 | 108.02 | 70.2 | 9.272 | 43.427 | 20.355 | 28.603 |
| 2,5-Dimethyl-hexane | 0.35683 | 105.72 | 68.6 | 9.051 | 39.877 | 19.745 | 27.872 |
| 2,2,4-Trimethyl-pentane | 0.30537 | 104.09 | 64.87 | 8.402 | 45.874 | 21.072 | 29.7 |
| 4-Methyl-heptane | 0.371504 | 109.32 | 70.91 | 9.483 | 38.846 | 19.282 | 27.414 |
| 3,3-Dimethyl-hexane | 0.322596 | 104.74 | 68.5 | 8.973 | 46.417 | 20.987 | 29.462 |
| 3-Methyl-3-ethyl-pentane | 0.306899 | 101.48 | 69.3 | 9.081 | 46.797 | 21.049 | 29.457 |
| 2,2,3-Trimethyl-pentane | 0.300816 | 101.31 | 67.3 | 8.826 | 51.003 | 22.021 | 30.881 |
| 3,4-Dimethyl-hexane | 0.340345 | 106.59 | 70.2 | 9.316 | 45.268 | 21.446 | 28.989 |
| 2,3,3-Trimethyl-pentane | 0.293177 | 102.06 | 68.1 | 8.897 | 52.748 | 22.349 | 31.282 |
| Octane | 0.397898 | 111.67 | 73.19 | 9.915 | 34.182 | 18.249 | 25.757 |
| 2,3,4-Trimethyl-pentane | 0.317422 | 102.39 | 68.37 | 9.014 | 48.144 | 21.404 | 30.014 |

TABLE 1. The values of Acentric factor, Entropy, Enthalpy of vaporization (HVAP), Standard enthalpy of vaporization (DHVAP), SOve, Nve, and MPI_{ve} for octane isomers.

| Index | Acentric Factor | Entropy | HVAP | DHVAP |
|------------|-----------------|--------------|--------------|--------------|
| SO_{ve} | -0.975070878 | -0.9213159 | -0.715882049 | -0.799150905 |
| N_{ve} | -0.959409353 | -0.91200543 | -0.735557078 | -0.803513767 |
| MPI_{ve} | -0.971177506 | -0.935705946 | -0.789606029 | -0.847466686 |

TABLE 2. The correlation between ve-degree topological indices and many physicochemical properties of octane isomers.

Corollary 3.1. Let $K_{n,n}$ be a complete bipartite graph $(n \ge 4)$. Then

- i. $SO_{ve}(K_{n,n}) = \sqrt{2}n^4$
- ii. $N_{ve}(K_{n,n}) = \sqrt{2}n^3$ iii. $MPI_{ev}(K_{n,n}) = n^3$.

Corollary 3.2. Let $K_{1,n-1}$ be a star graph $(n \ge 4)$. Then

i. $SO_{ve}(K_{1,n-1}) = \sqrt{2}(n-1)^2$

ii. $N_{ve}(K_{1,n-1}) = (n-1)\sqrt{2(n-1)}$ iii. $MPI_{ev}(K_{1,n-1}) = (n-1)(\sqrt{n-1}).$

Proposition 3.2. Let C_n be a cycle graph with $n \ge 4$. Then

- i. $SO_{ve}(C_n) = 4n\sqrt{2}$ ii. $N_{ve}(C_n) = 2n\sqrt{2}$
- iii. $MPI_{ev}(C_n) = 4n.$

Proof. Let C_n be a cycle graph with $n \geq 4$ vertices. Clearly, every vertex of $v \in V(C_n)$ has 4 different edges that incident to any vertex form N[v].

i.

$$SO_{ve}(C_n) = \sum_{uw \in E} \sqrt{d_{ve}^2(u) + d_{ve}^2(w)} = \sum_{uw \in E} \sqrt{4^2 + 4^2} = 4n\sqrt{2}$$

ii.

$$N_{ve}(C_n) = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)} = \sum_{uw \in E} \sqrt{4+4} = 2n\sqrt{2}$$

iii.

$$MPI_{ve}(C_n) = \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right) = \sum_{uw \in E} \left(\sqrt{4} + \sqrt{4} \right) = 4n$$

Proposition 3.3. Let K_n be a complete graph with $n(\geq 4)$ vertics. Then

i.
$$SO_{ve}(K_n) = \frac{n^2(n-1)^2}{2\sqrt{2}}$$

ii. $N_{ve}(K_n) = \frac{(n(n-1))^{3/2}}{2}$
iii. $MPI_{ev}(K_n) = \frac{(n(n-1))^{3/2}}{\sqrt{2}}$.

Proof. Let K_n be a complete graph with n vertices. Clearly, every vertex of $v \in V(k_n)$ has n(n-1)/2different edges that incident to any vertex form N[v].

$$SO_{ve}(K_n) = \sum_{uw \in E} \sqrt{d_{ve}^2(u) + d_{ve}^2(w)} = \sum_{uw \in E} \sqrt{[n(n-1)/2]^2 + [n(n-1)/2]^2} = \frac{n^2(n-1)^2}{2\sqrt{2}}$$
ii.

 $N_{ve}(K_n) = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)} = \sum_{uw \in E} \sqrt{[n(n-1)/2] + [n(n-1)/2]} = \frac{(n(n-1))^{3/2}}{2}$

iii.

i.

ii.

$$MPI_{ve}(K_n) = \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right) = \sum_{uw \in E} \left(\sqrt{[n(n-1)/2]} + \sqrt{[n(n-1)/2]} \right)$$
$$= \frac{(n(n-1))^{3/2}}{\sqrt{2}}$$

Proposition 3.4. For the path graph $P_n(n \ge 5)$, $d_{ve}(v_1) = d_{ve}(v_n) = 2$, $d_{ve}(v_2) = d_{ve}(v_{n-1}) = 3$, and remaining vertices has the ve-degree 4. Then,

$$SO_{ve}(P_n) = 2\sqrt{13}(n-1) + 10(n-1) + 4\sqrt{2}(n-1)(n-5)$$
$$N_{ve}(P_n) = 2\sqrt{5}(n-1) + 2(n-1)\sqrt{7} + 2\sqrt{2}(n-1)(n-5)$$

iii.

$$MPI_{ve}(P_n) = (n-1)[2\sqrt{2} + 4\sqrt{3} + 4(n-5) + 4]$$

Proof. With the definitions and ve-degree of each vertex one can easily arrive the results.

4. Bounds

In this section, we found the bounds for SO_{ve} and N_{ve} in terms of MPI_{ve} , $M'_{\beta ve}(G)$, and $F_{ve}(G)$.

Lemma 4.1. For any positive numbers α and β ,

$$\frac{1}{\sqrt{2}}(\alpha+\beta) \le \sqrt{\alpha^2+\beta^2} \le \alpha+\beta.$$

Equality on the left-hand side holds if and only if $\alpha = \beta$.

Theorem 4.1. For any non-trivial connected graph G,

$$\frac{1}{\sqrt{2}}M'_{\beta ve}(G) \le SO_{ve}(G) \le M'_{\beta ve}(G).$$

Proof. With help of definitions and lemma 4.1, we arrive the result. **Lemma 4.2.** Let α and β be any non-negetive real numbers. Then

$$\sqrt{\alpha + \beta} \ge \frac{1}{\sqrt{2}} \left(\sqrt{\alpha} + \sqrt{\beta} \right)$$

Theorem 4.2. Let G be a connected graph of order n and size m. Then

$$\frac{1}{\sqrt{2}}MPI_{ve}(G) \le N_{ve}(G) \le MPI_{ve}(G).$$

Proof. If $\alpha = d_{ve}(u)$ and $\beta = d_{ve}(w)$ in Lemma 4.2, then we get

$$\sqrt{d_{ve}(u) + d_{ve}(w)} \ge \frac{1}{\sqrt{2}} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right)$$

By the definition of *ve*-Nirmala index, we have

$$N_{ve}(G) = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)} \ge \frac{1}{\sqrt{2}} \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right) = \frac{1}{\sqrt{2}} MPI_{ve}(G)$$

By the definition of *ve*-Nirmala index, we have

$$N_{ve}(G) = \sum_{uw \in E} \sqrt{d_{ve}(u) + d_{ve}(w)}$$
$$= \sum_{uw \in E} \left[\left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right)^2 - 2\sqrt{d_{ve}(u)d_{ve}(w)} \right]^{1/2}$$
$$\leq \sum_{uw \in E} \left(\sqrt{d_{ve}(u)} + \sqrt{d_{ve}(w)} \right) = MPI_{ve}(G)$$

Theorem 4.3. Let G be a connected graph with m edges. Then

$$SO_{ve}(G) \le \sqrt{mF_{ve}(G)}.$$

Proof. Using the Cauchy-Schwarz inequity, we get

$$\left(\sum_{uw\in E} d_{ve}(u)^2 + d_{ve}(w)^2\right)^2 \le \sum_{uw\in E} 1 \sum_{uw\in E} \left(d_{ve}(u)^2 + d_{ve}(w)^2\right) = mF_{ve}(G)$$

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CONCLUSION

In this study, we have computed SO_{ve} , N_{ve} , and MPI_{ve} for Standard graphs, and found the bounds for SO_{ve} and N_{ve} in terms of MPI_{ve} , $M'_{\beta ve}(G)$, and $F_{ve}(G)$. The predictive ability of the *ve*-degree index is greater and also has a better correlation than classic degree-based indices. Also, the predictive ability of newly defined *ve*-degree indices has been tested on some physicochemical properties of octanes.

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